

2. V. I. Bol'shakova, V. A. Khan, Zh. V. Dubovenko, É. N. Shmidt, and B. A. Pentegova, *Khim. Prir. Soedin.*, 340 (1980).
3. V. A. Khan, Zh. V. Dubovenko, and V. A. Pentegova, *Khim. Prir. Soedin.*, 111 (1976).
4. É. N. Shmidt and V. A. Pentegova, *Khim. Prir. Soedin.*, 675 (1974).

#### SYNTHESIS OF DEUTERATED (-)-LIMONENE

É. F. Buinova, T. I. Pekhk,  
V. I. Lysenkov, and T. L. Senko

UDC 547.596.7

Studying the isomerization transformations of limonene, we have synthesized for the first time  $[3,3\text{-}^2\text{H}_1]$ - and  $[3\text{-}^2\text{H}_1]$ limonenes (ratio 3:1) by the dehydration of a mixture of deuterated stereoisomeric dihydrocarveols obtained by the reduction of deuterated dihydrocarbone.

According to our results (GLC and  $^{13}\text{C}$  NMR spectrum), (-)-dihydrocarvone obtained from carvone [1] contains the cis and trans isomers in a ratio of 1:5.

The deuteration of the dihydrocarvone was performed by the method of Reich et al. [2]. A reaction product was obtained with a yield of 81% containing, according to GLC, the cis and trans isomers in a ratio of 1:5, bp 70-72°C (2 mm),  $[\alpha]_{\text{D}}^{20} - 16^\circ$ ,  $n_{\text{D}}^{20} 1.4708$ . IR spectrum ( $\nu$ , in a thin layer),  $\text{cm}^{-1}$ : 895, 1640 ( $\text{CH}_2=\text{CH}$ ), 1715 ( $\text{C}=\text{O}$ ), 2140, 2235, 2375 (C-D). Mass spectrum:  $M^+$  155, 154, 152; isotopic composition: 81% of  $[1,3,3\text{-}^2\text{H}_3]$ dihydrocarvone (I), 15% of  $[3,3\text{-}^2\text{H}_2]$ dihydrocarvone, and 4% of undeuterated dihydrocarvone.  $^{13}\text{C}$  NMR spectrum of the trans isomer (I) ( $\text{CDCl}_3$ , 0 - TMS;  $\delta$ , isotope shift, ppm) 44.2, -0.5 ( $\text{C}^1$ ); 46.2, -0.7 ( $\text{C}^3$ ).

A mixture of 38 g of the deuterated dihydrocarvone, 10 g of sodium tetrahydroborate, and 300 ml of dioxane was stirred at room temperature for 10 h. After the usual working up and vacuum distillation, 32 g (85%) of a product was obtained which, according to GLC, contained the four stereoisomeric dihydrocarveols in a ratio of 1:4:10:15, bp 83-85°C (3-5 mm),  $[\alpha]_{\text{D}}^{20} + 3^\circ$ ,  $d_4^{20} 0.9415$ ,  $n_{\text{D}}^{20} 1.4781$ . IR spectrum ( $\nu$ , in a thin layer,  $\text{cm}^{-1}$ ): 890, 1640 ( $\text{CH}_2=\text{CH}$ ), 2120, 2200 (C-D), 3360 (O-H).

At 110°C, 32 g of dihydrocarveol was added to a solution of 150 g of p-toluenesulfonyl chloride in 740 ml of pyridine, and the mixture was stirred for 5 h. After the usual working up of the reaction mixture and vacuum distillation, 9.0 g (32%) of a hydrocarbon fraction (bp 51-52°C/5-6 mm,  $[\alpha]_{\text{D}}^{20} - 106^\circ$ ,  $d_4^{20} 0.8546$ ,  $n_{\text{D}}^{20} 1.4716$ ) was isolated, which, according to GLC, contained 96% of limonene, 3.4% of trans-isolimonene, and 0.6% of an unknown hydrocarbon, probably cis-isolimonene. IR spectrum ( $\nu$ , in a thin layer),  $\text{cm}^{-1}$ : 870, 1640 ( $\text{CH}_2=\text{CH}$ ), 2059, 2155 (C-D). Mass spectrum:  $M^+$  138, 137, 136; isotopic composition: 70% of  $[3,3\text{-}^2\text{H}_2]$  limonene (II), 21.5% of  $[3\text{-}^2\text{H}_1]$ limonene, and 8.5% of undeuterated limonene.  $^{13}\text{C}$  spectrum of the limonene (II) ( $\text{C}_6\text{D}_6$ , 0 - TMS;  $\delta$ , isotopic shift, ppm): 33.39, 0.110 ( $\text{C}^1$ ), 122.11-0.126 ( $\text{C}^2$ ), 31.98 -0.280 ( $\text{C}^3$ ), 41.40, -1.180 ( $\text{C}^4$ ) 28.34 -0.039 ( $\text{C}^5$ ), 30.98, 0.004 ( $\text{C}^6$ ), 23.57, 0.006 ( $\text{C}^7$ ), 149.67, -0.006 ( $\text{C}^8$ ), 108.90, 0.019 ( $\text{C}^9$ ), 20.85, 0.008 ( $\text{C}^{10}$ ).

#### LITERATURE CITED

1. T. G. Halsall, D. W. Theobald, and K. B. Walshaw, *J. Chem. Soc.*, 1029 (1964).
2. H. J. Reich, M. Jautelat, M. T. Messe, F. J. Weigert, and J. D. Roberts, *J. Am. Chem. Soc.*, 91, 7454 (1969).

---

Institute of Physical Organic Chemistry of the Academy of Sciences of the Belorussian SSR, Minsk, and Institute of Chemical and Biological Physics of the Academy of Sciences of the Estonian SSR, Tallin. Translated from *Khimiya Prirodnikh Soedinenii*, No. 1, p. 111, January-February, 1983. Original article submitted July 7, 1982.